

Electrochemical Li recovery strategy from Li-ion battery black mass

Pier Giorgio Schiavi

Department of chemistry

Sapienza University of Rome

piergiorgio.schiavi@uniroma1.it

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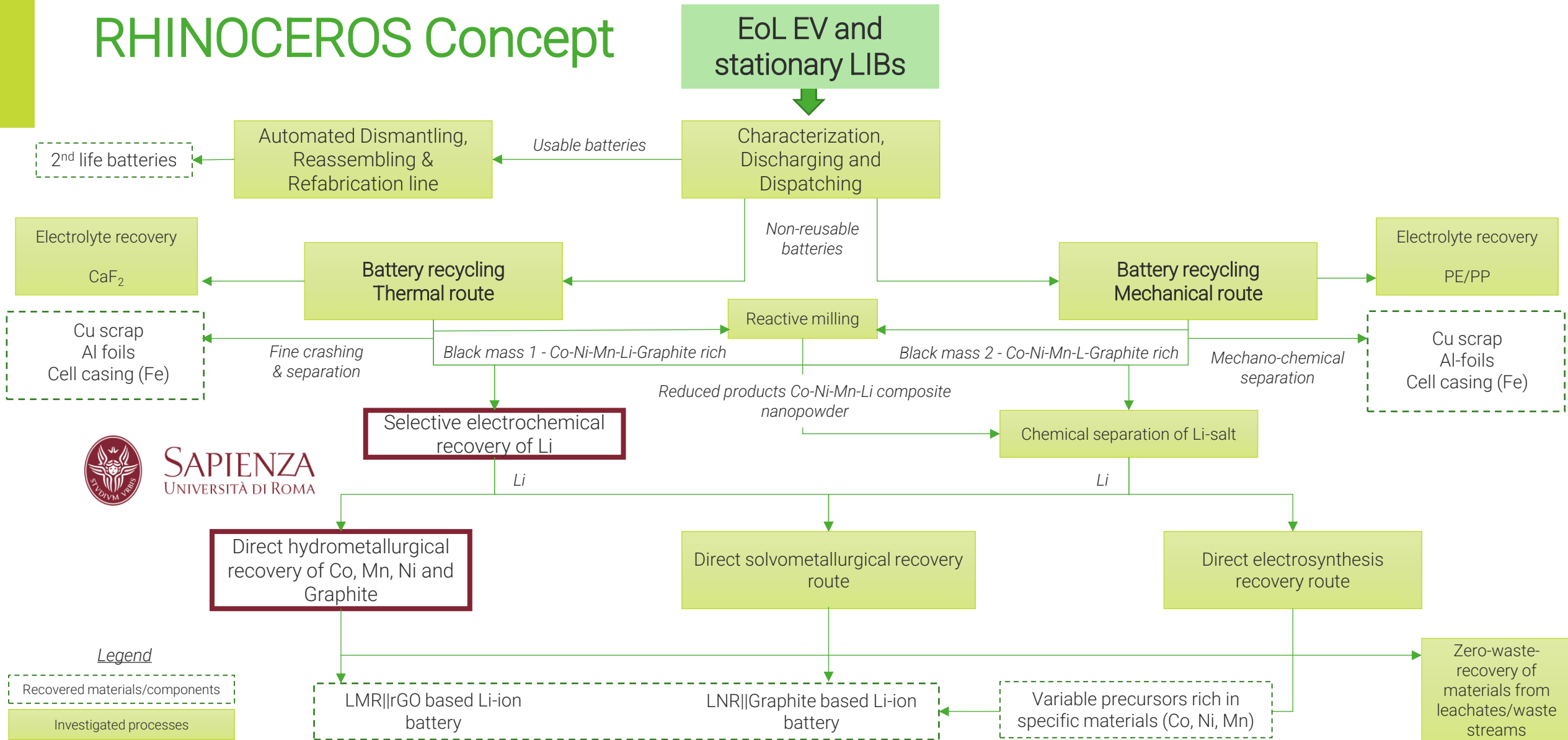
RHINOCEROS Objective

- The objective of Rhinoceros is to develop, improve, and demonstrate, in an industrially relevant environment, an economically and environmentally viable route for re-using, and recycling End-of-Life (EoL) Electric Vehicles (EV) and stationary energy storage Lithium-Ion Batteries (LIBs).

 **16** partners from **9** countries



RHINOCEROS Concept



Legend
 Recovered materials/components
 Investigated processes

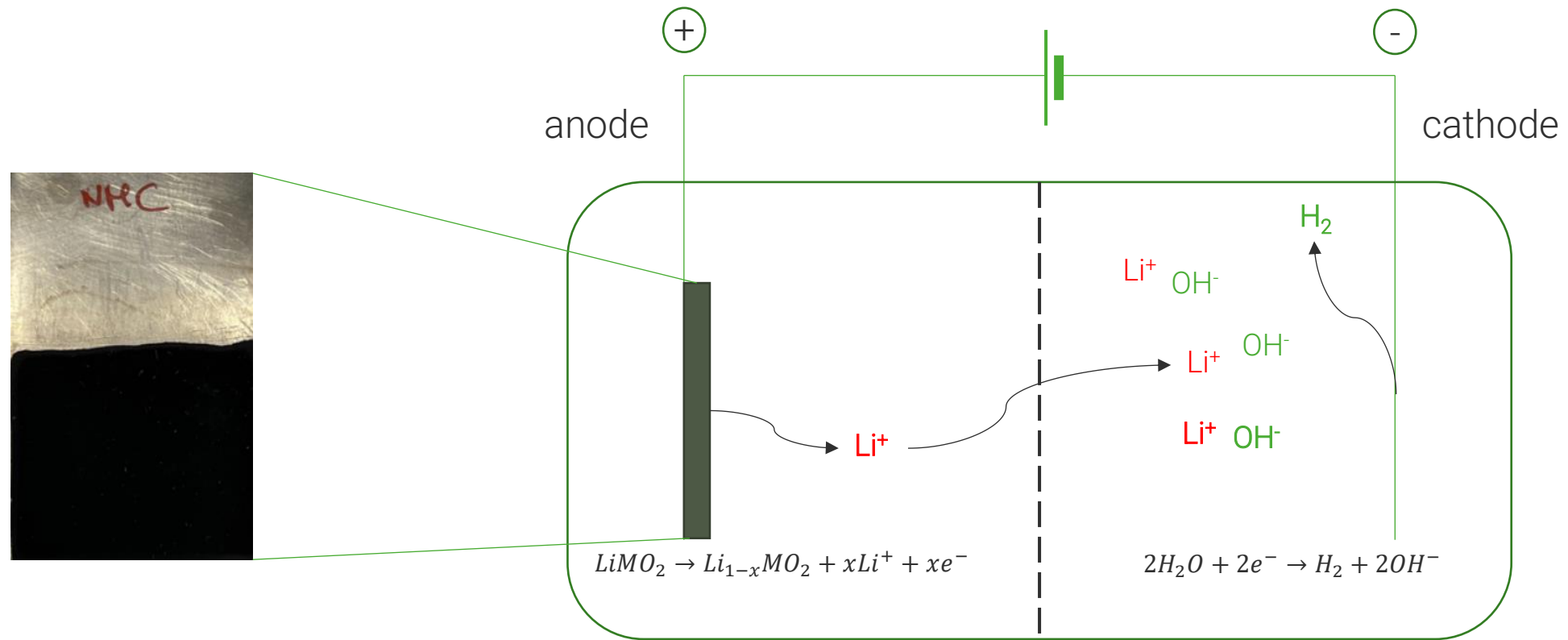


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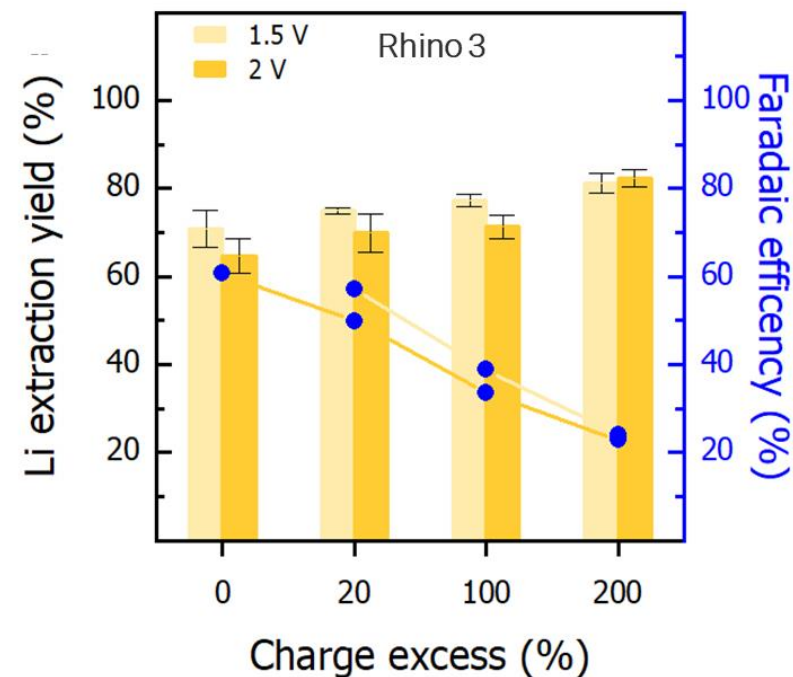
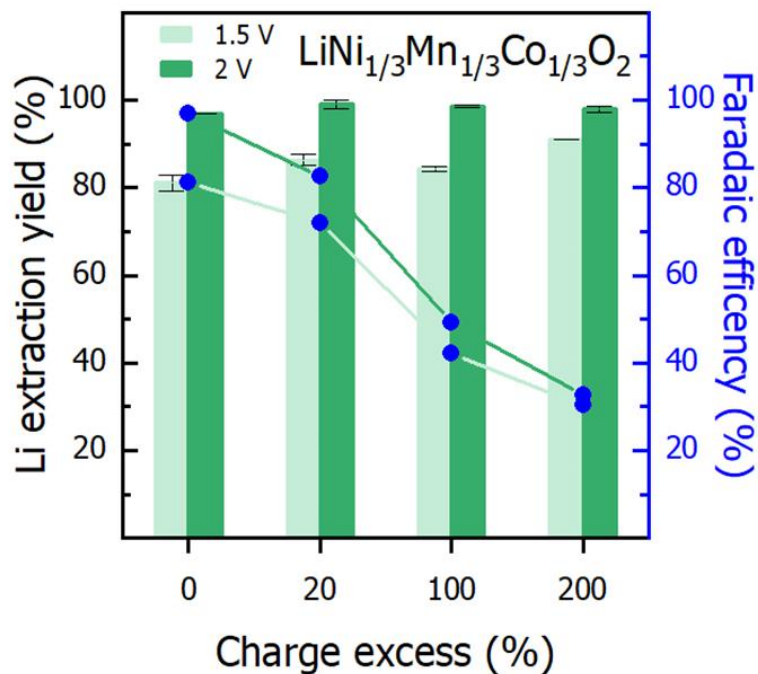
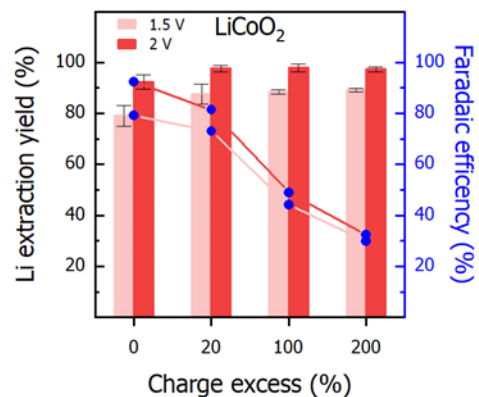
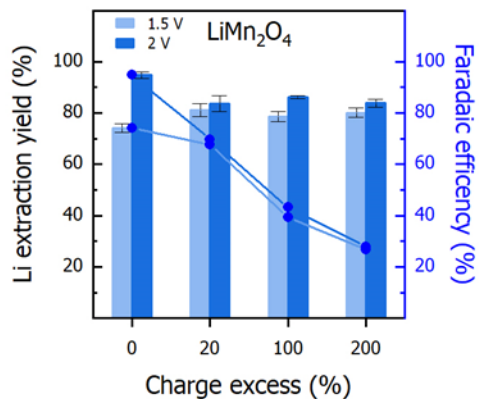
Electrochemical recovery of Lithium: concept

- This activity aims to develop an electrochemical process for selective extraction of Li from the EoL LIB black mass, mimicking the charging process of a standard LIB.

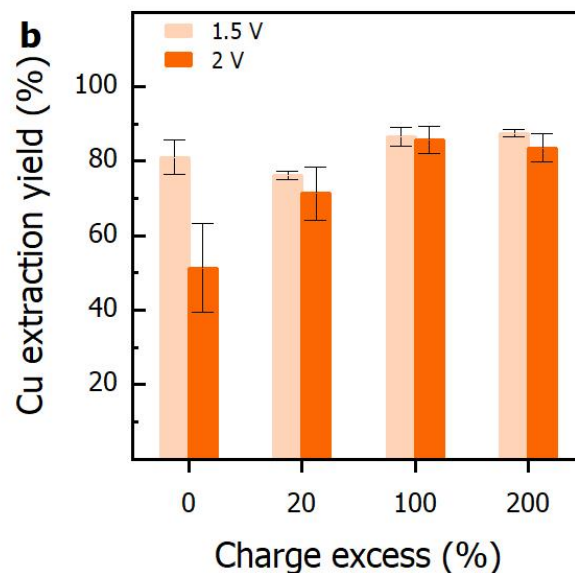
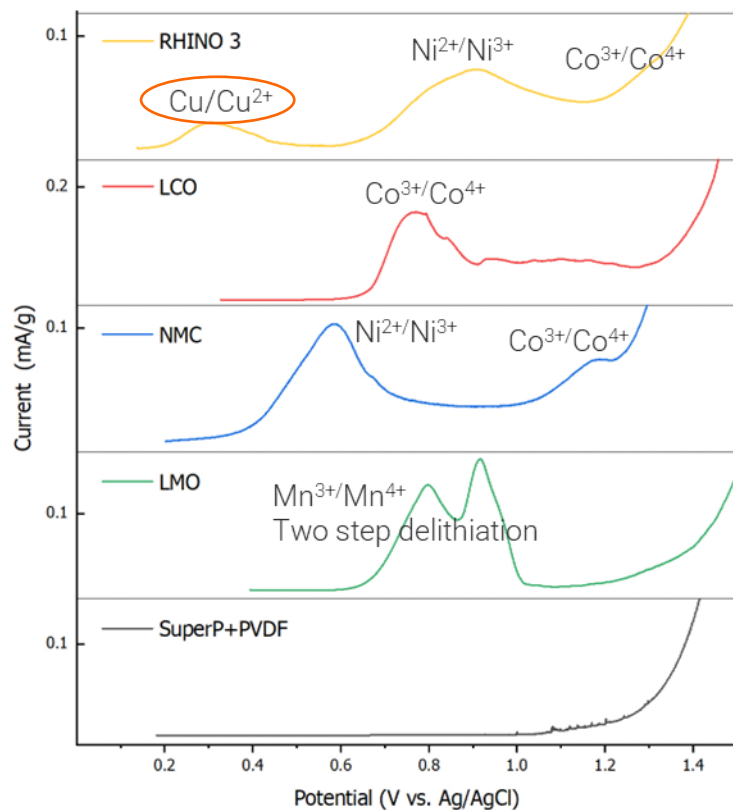


Potentiostatic delithiation

- Li extraction yields of up to 82% were achieved with electrodic powders from EoL LIBs, while commercial electrodic powders reached 99%.



Role of impurities: Cu

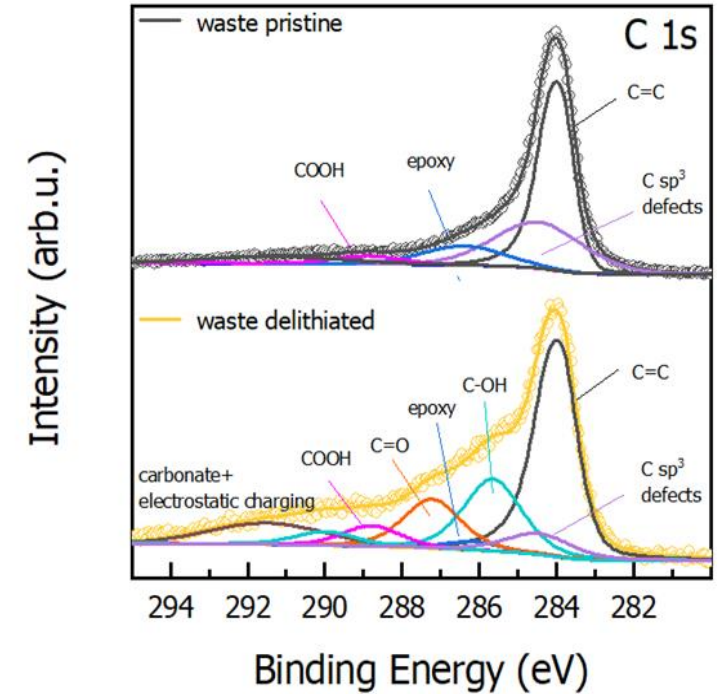
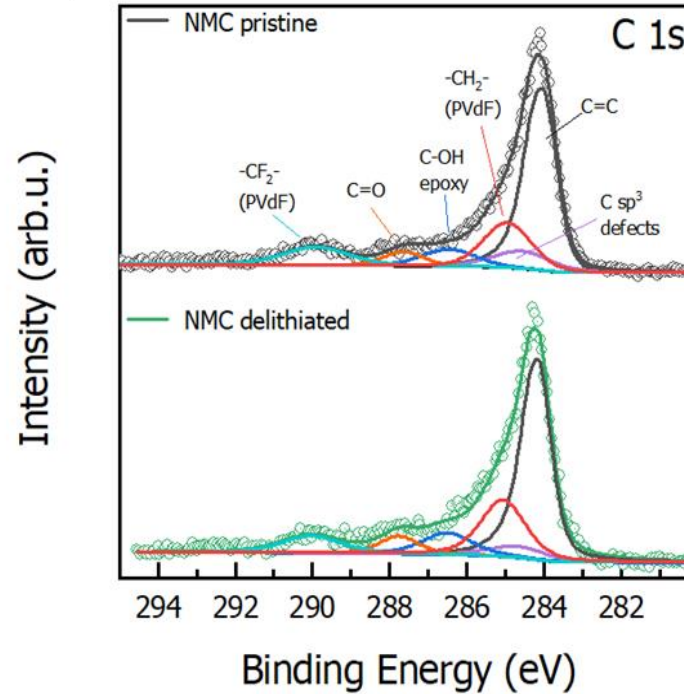
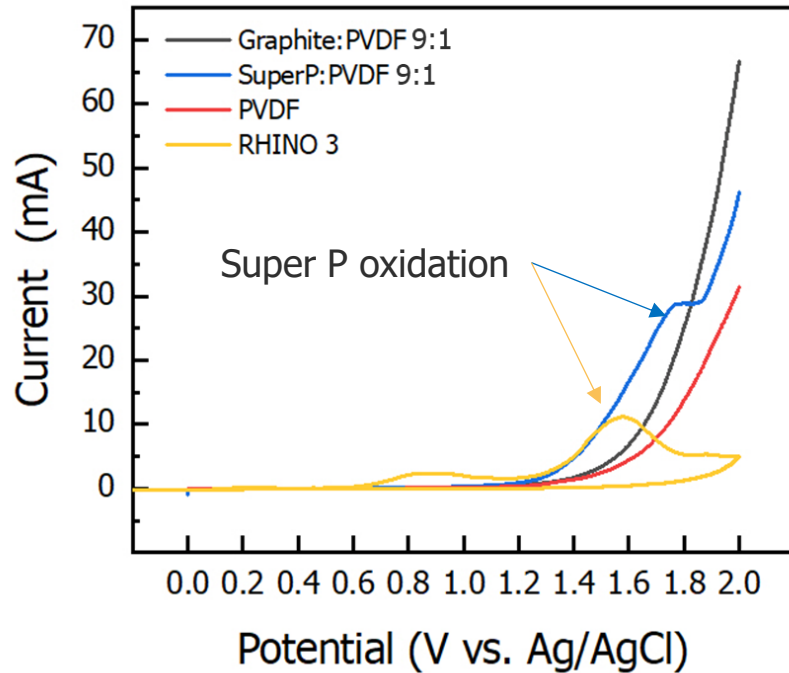


RHINO-3 – Black mass		
	average [mg/g]	st.dev [mg/g]
Co	75	3
Ni	92	1
Mn	88	6
Li	31	1
Al	6	5
Cu	13	1
Fe	1	0

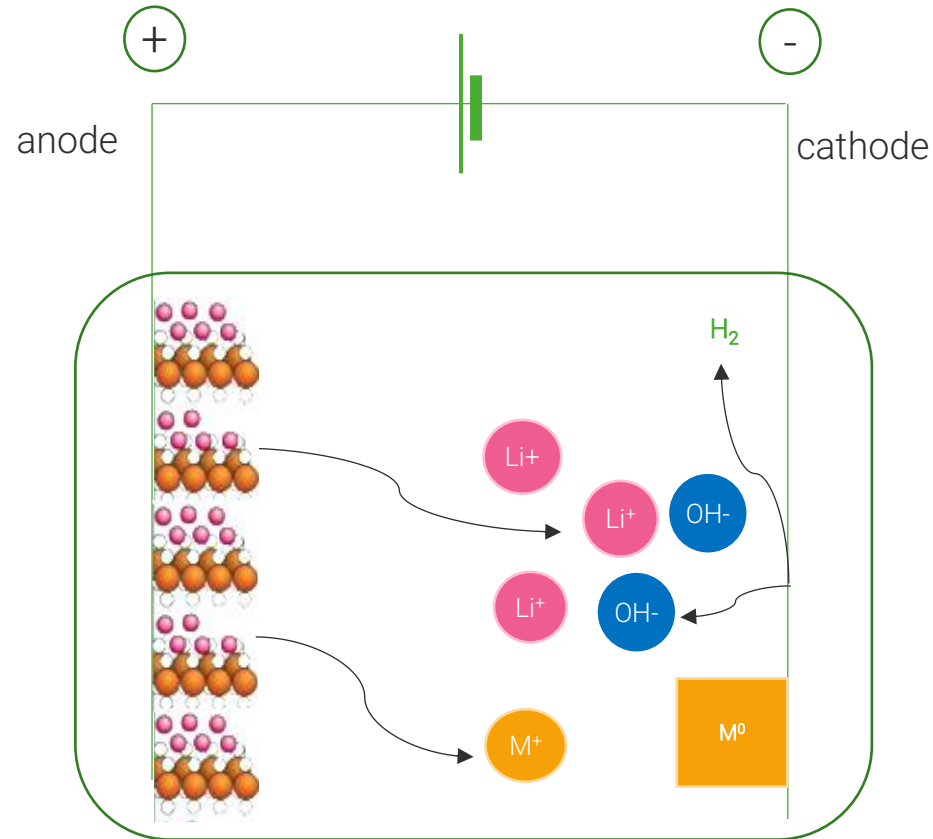
Theoretical charge for the complete oxidation of Cu ~ 40 C/g

- corresponding to $\sim 9\%$ of theoretical charge imposed for the oxidation of NMC in R3 (493 C/g)
- corresponding to $\sim 3\%$ of total charge in the case of 200% charge excess respect theoretical charge for NMC oxidation (1480 C/g)

Role of impurities: graphite and super P conductive carbon



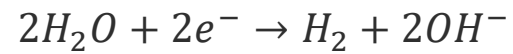
Li selectivity towards Ni, Mn and Co



Metal	Extraction %
Co	0.04±0.005
Ni	0.07±0.009
Mn	0.02±0.002

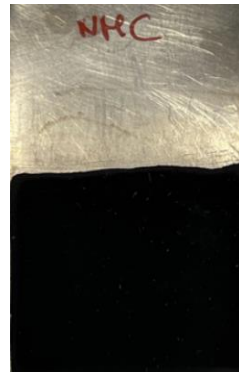
Experimental Procedure:

- Delithiation of Rhino 3 at 2V with a 200% charge excess (6 replicates).
- ICP-OES analysis of the electrolytes (no metals detected).
- Digestion of counter electrodes used during delithiation, followed by analysis with ICP-OES

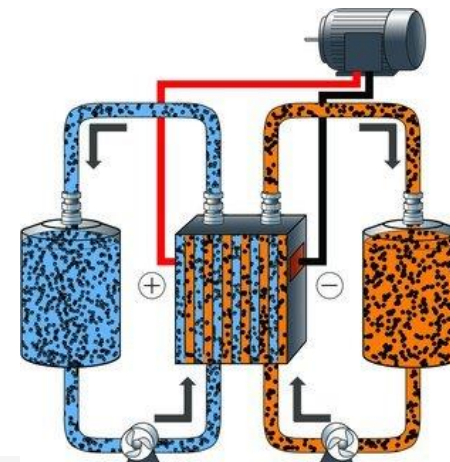


Conclusions

- **Efficiency:** Rhino 3 displays lower extraction yields (82%) and faradaic efficiency to respect commercial cathode materials (close to 100%). Super P oxidation could explain the lower Li extraction
- **Selectivity:** Ni, Mn and Co are not extracted
- **Scaling up:** evaluating an alternative approach that enables the treatment of larger quantities of powder without replicating the manufacturing process of LIB electrodes!

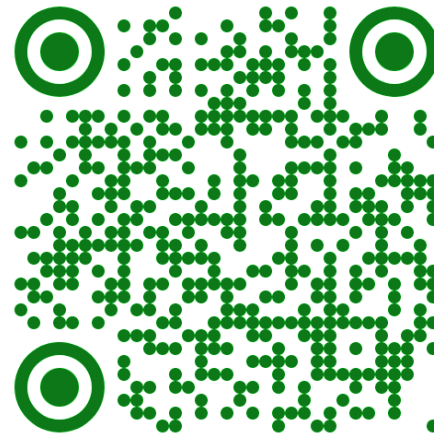


Slurry electrolysis

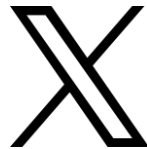


Thanks for your attention!

piergioorgio.schiavi@uniroma1.it



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